



One Day National Conference on A Glimpse of Differential Equations in Science and Engineering



2019-20

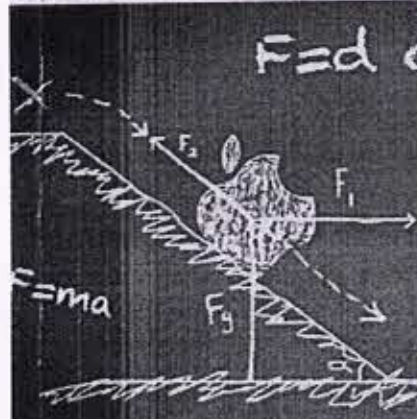
$$u(t) \quad \frac{d^2y}{dt^2} + \dots$$

$$\frac{2}{\sin^2 x} + 1) \dots$$

$$A = \pi r^2$$

$$= \frac{-b \pm \sqrt{b^2 - 4ac}}{2a}$$

$$\binom{n}{r} x^r a^{n-r}$$



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APPLICATION OF DIFFERENTIAL EQUATION IN PHYSICS

Kunale R.A.

Kai. Rasika Mahavidyalaya Deoni, Dist Latur, Maharashtra
Email: kunalerenuka9@gmail.com

ABSTRACT:

Differential equation gives us information about how a one of the quantities (i.e. dependent variable) changes with respect to (independent variable). The different variable used in physics are time, position, distance, acceleration, temperature, space etc. In ordinary differential equation of variation of position with respect to time contains the x , t , dx/dt , d^2x/dt^2 and several higher order derivatives. There are different methods to solve differential equation, using that solution we get the information about how quantity changes with respect to expected variable. In physics differential equation are used to represent periodic motion for instance in simple harmonic motion to represent variation of position with respect to time i.e. dx/dt is used.

INTRODUCTION:

An equation involving derivatives or differentiation of one or more dependent variable with respect to one or more independent variable is called differential equation [1-2]. Almost all the theories of physics have expressed physical laws by using differential equation. A law is set of rules followed by particular system. Such a rules are observed by repeated experimentation. For example, Newton's law are set of rules followed by a system possessing motion. The second law of motion is expressed in the form of second order ordinary differential equation $F(x) = m d^2x/dt^2$, here x is distance of a point mass at any instant 't' from point 'o' taken as origin on a straight line [3,4]. The equation of this kind is called differential equation because this equation contains d^2x/dt^2 . This is a second order differential equation. The set of rules for making a law are fixed and system follows these rules under different circumstances. However, the laws can break-down in some situations, and therefore a law is true only for a subset of conditions. For example, all macroscopic systems using Newton's laws of motion one can find the exact position and momentum of an object at any instant of time whereas for quantum sized systems one cannot determine momentum and position values at any instant of time with full certainty. Another example is that of ohms law, for certain systems such as copper wire one can find linear relationship between current and voltage, whereas for systems such as transistors such relationship breakdowns. According to classical physics, all systems change continuously as time passes. A physical law expresses a permanent relationship between the state of the system at the present moment and its state immediately following that moment. Therefore, a law must define the connection between successive space-time points. As we can see in equation, Newton's second law of motion defines a relationship that connects the velocity of the body at a present space-time point with the velocity at a point in the immediate neighbourhood of


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the first space-time point. A majority of physical processes change continuously, and therefore a law must express this continuity of change in a causal chain. In a continuous process, the sequence of states assumed by a physical system forms a continuous chain, which means that the change in state of the system over an infinitesimal period of time is itself infinitesimal. The laws controlling such processes must therefore exhibit continuity of change.

APPLICATION IN WAVE MOTION OF STRING:

When we have a function $y(t)$, we can readily define dy/dx as the slope of the plot $y(x)$. But now consider $y(x,t)$. In our example, this will be the displacement y of a point on a string as a function of position on the string x , and time t . So we can now think of two different derivatives. We write them differently. What we shall do here is to solve the wave equation, the equation of motion for a wave in a string[5].

$\partial y/\partial x$, think of this as dy/dx at a given constant time, t . Imagine taking a photograph (time is constant): in the image at time t , this is the slope of the $y(x)$ shape at the instant of the photograph.

$\partial y/\partial t$. Think of this as dy/dt at a given position, x . This is just the velocity in the y direction at a particular point x on the string. (Not the velocity of the wave, by the way).

Let's take a standard example. A travelling sine wave with amplitude A , frequency $f = 2\pi\omega$ and wavelength $\lambda = 2\pi/k$ has the equation

$$y = A \sin(kx - \omega t),$$

$\partial y/\partial x = kA \cos(kx - \omega t)$, which is the slope of the string at position x and time t ,

$\partial y/\partial t = -\omega A \cos(kx - \omega t)$, which is the velocity of a point on the string at x and t .

$\partial^2 y/\partial x^2 = -k^2 A \sin(kx - \omega t)$, which is the rate of change in the slope of the string, as x varies,

And

$\partial^2 y/\partial t^2 = -\omega^2 A \sin(kx - \omega t)$, which is the acceleration of a point on the string.

These have important physical significance: the first one is determining the curvature of the string. If $\partial^2 y/\partial x^2 = 0$, then the slope is constant, so it is straight. That means that the tension T acts in opposite directions at opposite ends, giving no net force. If a segment is curved, however ($\partial^2 y/\partial x^2 \neq 0$), it has a force acting on it. For constant curvature over a small length L , the net force is proportional to L as shown in Fig. 1.

We know the acceleration so we can apply Newton's second law. The mass of the segment is μL , where μ is the mass per unit length μ . Writing Newton's law as $a = F/m$ gives:

$$\partial^2 y/\partial t^2 = (T/\mu)\partial^2 y/\partial x^2$$

Looking back at our expressions for the two second derivatives, we see that they our original function $y = A \sin(kx - \omega t)$ is a solution to the wave equation, provided that $T/\mu = \omega^2/k^2$. We also know in wave, that ω/k is the wave speed, v . Which finally relates the wave speed to the physical properties T and μ of the string:

$$v = \sqrt{(T/\mu)}.$$

The wave speed is greater if the string is stretched more tightly, and less if the string has a high mass per unit length.



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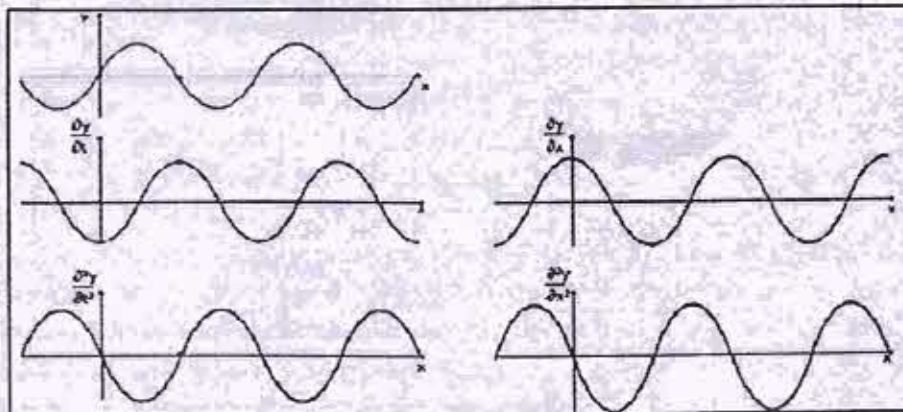


Fig.1 Variation of y , dy/dt , d^2y/dt^2 , dy/dx , d^2y/dx^2 with position.

CONCLUSION:

Differential equations are used for formulation of various laws of physics for example Newton's law, wave equation of string, radioactive decay etc. Differential equations are the foundation for understanding many of the most important processes and phenomenon in nature. Technological progress in some area occurred with the solution to differential equation to guide and provide confident solution and insight.

REFERENCES:

1. Simplified course in differential equation, M. D. Raisinghania, H. C. Saxena, H.K. Das, S. chand and comp. private ltd.
2. Advanced engineering mathematics, H. K. Das.S. chand and comp. private ltd.
3. Differential equation and their application, H.T.H. Piaggio, CBS publisher.
4. Why are differential equations used for expressing the laws of physics, Shabnam Siddiqui.
5. Physics UNSW, differential equation.

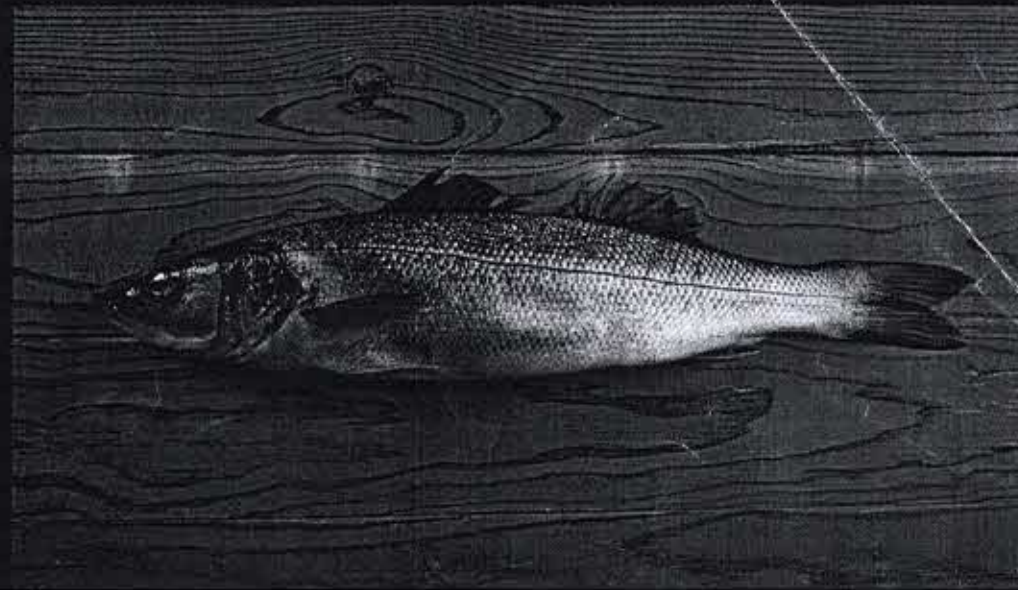


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Indian major carps, riverine fishes which normally breed in rivers during the months from June to August. They normally do not breed in captivity. There is need to induce Indian major carps to breed in confined waters and thereby ensure a dependable source of quality seed to enhance the fish production and development of fishery.

The breeding of fish in the controlled condition is very important for the production of quality fish seed in sufficient quantity. Hence, the present research works are carried out during June - August 2009 and June-August 2010. At fish breeding center, Jayakwadi, Paithan, Dist. Aurangabad in Maharashtra state, India.

A reference book on 'Induced Breeding Technique' has been written with the main objective to provide the knowledge for students, researchers, planners, fish breeders and hatchery managers. Authors are sure the book will be helpful to them.



Purushottam More
Ravindra Bhandare



Dr. Purushottam R. More is presently working as an Assistant Professor and Head, Department of Zoology, Kai. Rasika Mahavidyalaya, Deoni. District Latur (MS) India.



Dr. Ravindra Y. Bhandare is presently working as an Assistant Professor, Department of Zoology MGVS Arts, Science and Commerce College, Surgana District Nasik (MS) India.

Induced Breeding Technique

A Reference Book



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Kai. Rasika Mahavidyalaya, Deoni
Tq. Deoni Dist. Latur

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A reference book of 'Induced Breeding Technique' will be helpful to the students, researchers, planners, fish breeders and hatchery managers.

A REFERENCE BOOK OF

INDUCED BREEDING TECHNIQUE

Dr. Purushottam R. More

Head, Department of Zoology,

Kal. Rasika Mahavidyalaya, Deoni.

District Latur (MS) India.

Dr. Ravindra Y. Bhandare

Asst. Professor

Department of Zoology

MGV'S Arts, Science and Commerce

College, Surgana District Nasik (MS)

India.


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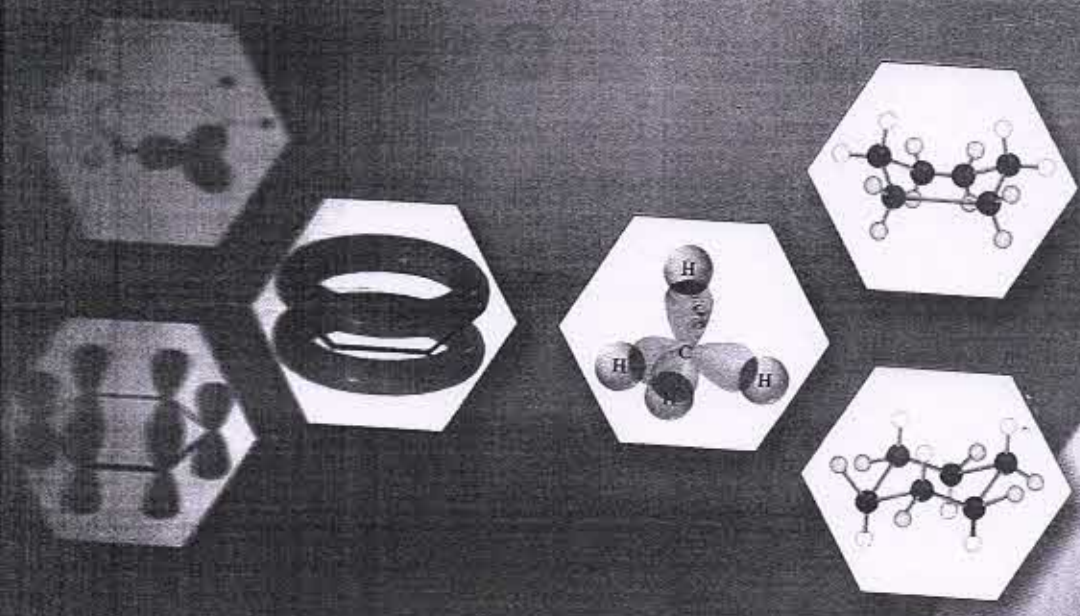
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A Text Book of Organic Chemistry

B.Sc. First Year
(Semester I and II)

Dr. Chandrashekhar Malba
Mr. Bhaskar Ankush ✓
Dr. Vijaykumar More



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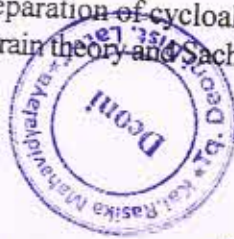


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In this book some basic programming approach of MATLAB has been given. following are contents which is focused in this book.

- Create Matrices and it's operations, plotting graphs using MATLAB.
- Plotting multiple graphs as well Transcendental graphs using MATLAB.
- Introduction to symbolic methods and solving problems.
- Working with Partial Differential Equations using mathematical software like MATLAB, maple etc.
- Solving problems in Numerical Analysis using mathematical software.



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Mathematical Programming

using MATLAB/Maple



Dr. Yogesh Manohar Muley,
 Head and Asst. Professor of Mathematics,
 Kai. Rasika Mahavidyalaya, Deoni, Dist. Latur, India.
 -have worked in Government Engineering College,
 Aurangabad (Full time on Contract).



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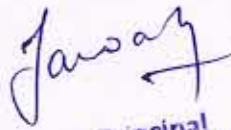
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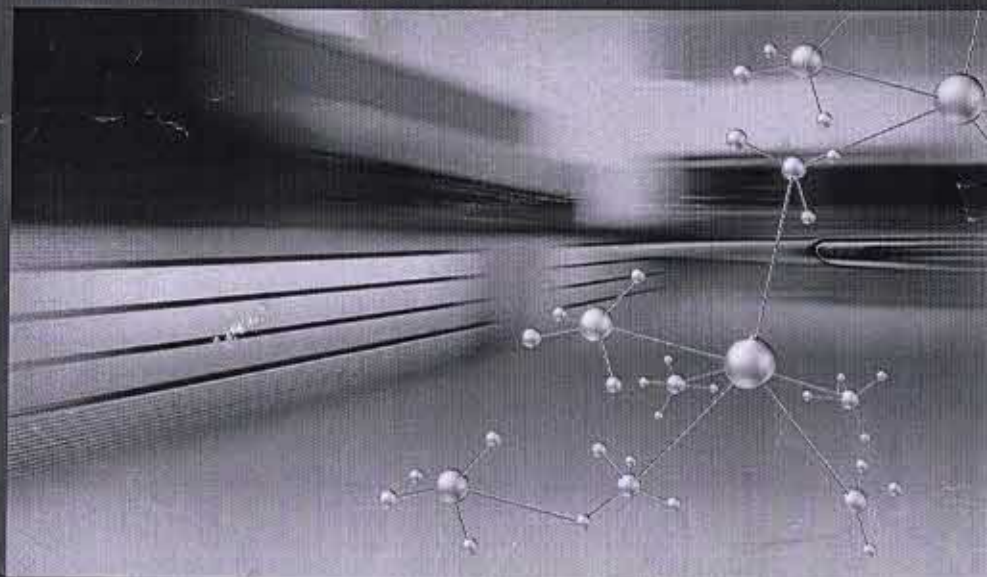


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This book explains the basic aspects of inorganic chemistry with a primary emphasis on facts; then uses the students growing factual knowledge as a foundation for discussing the important principles. This book contains separate chapters on periodic table and their properties, s-blocks elements, p-blocks elements, noble gas chemistry, acids and bases, chemical bonding and oxidation reduction. Each and every Concept has been explained with simple and suitable example. Sufficient number of multiple choice questions is given at the end of every topic. This book can be prove helpful to the graduation and under graduation students, researchers, academicians and all the scientific community interested in chemistry of inorganic materials.



Vijaykumar Shivdasrao More

Basic Aspects Of Inorganic Chemistry

For Degree Students



Dr. Vijaykumar Shivdasrao More,
Head Dept. of Chemistry,
Kai. Rasika Mahavidyalaya Deoni,
Dist-Latur, India-413519.
M.Sc., Ph.D.(University Of Salerno, Italy)
Post-Doc.(Kyoto Pharmaceuticals University, Japan)
He did 4 years research at CSIR-NCL, Pune (India) & 1
year at ITQB, New University of Lisbon, Portugal.
Published papers-20 & Books-06



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AA & GG AUDITS AND OTHER QUALITY MEASURES IN HIGHER EDUCATION INSTITUTIONS

18-12

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Dr. Arvind M. Nawale
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ACADEMIC AND ADMINISTRATIVE AUDIT: TOOLS FOR QUALITY IMPROVEMENT

Mr. Tenkale M.N

Assistant professor, Dept. of English
Kai.RasikaMahavidyalaya, Deoni.

Abstract: Today's age is known as the age of competition in every walks of life. Education is one of the field where the changes are taking place day by day. Academic and Administrative audits are to update and enhance the quality of educational institute. These are the tools by which an institute come up with their strength and weaknesses. These audits helps the institute to make improvement in their academic and administrative process. IQAC is an internal quality instrument established according to the guidelines of NAAC in the institute to plan and monitor the whole system. IQAC plays vital role in academic and administrative Audit.

Keywords: IQAC, Academic and Administrative audit

Academic and Administrative Audit: Tools for Quality improvement

Introduction -Education plays a vital role in the developments of any nation. It provides skilled professionals who leads their country on the top. Education is a growing sector where changes are taking place after every minutes. To compete with this change and to be update we need to revise the education system as per its need. Academic and administrative audit are the tools by which we can update and sustain our educational quality. Academic and administrative audit can be done internally and externally. Internal audit done by IQAC of the institution while external audit is done by university or by private agencies. In some state it is conducted by state agencies. In Gujarat knowledge construction of Gujrat has developed a very good model of academic and administrative audit.

Academic Audit - B.L. Gupta defines 'Academic audit is a systematic and scientific process of designing, implementing, monitoring and reviewing the quality of academic systems' It is related to quality assurance and enhancing the quality of academic activities in HIE.

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Administrative Audit -M. Rajendra defines administrative audit ' A method of assessing the efficiency and effectiveness of the operating system of administrative procedures, policies, decision making authorities and functionaries , strategies, process, feedback, control mechanism and so on' The AA would certainly make the functionaries to ascertain the strength and weakness of the operating system in general and pin out the areas in particular, and to ascertain where the function is stagnated and affected, and where special attention is required along with man and material resources" (P. 54)


Objectives of Academic and administrative audit

- It makes awareness to the institution of its strengths and weaknesses and suggests the mechanism for improvement.
- It helps to identify the problems in the present administrative process and find out the opportunities of improvement in the system.
- It helps to continuous improvement and sustain the quality.
- To setup the academic and administrative standards.
- To encourage the faculty to revise their educational quality methods.

Purpose of academic and administrative audit - The purpose of the academic and administrative audit is to evaluate and examine the educational institution and suggests the institute their achievements and pitfalls for further improvement. After the assessment of the institution the committee gives suggestion on the following points.

1. Availability of teaching and non-teaching faculty.
2. Infrastructural facilities available for carrying out academic and administrative activities.
3. Efforts taken for curricular development.
4. Teacher quality.
5. Teaching methods adopted and use of ICT in teaching, learning process.
6. Feedback mechanism used for assessing the performance of teachers by students and for curricular development.
7. Faculty development programs implemented by the department.
8. Strengths, Weaknesses. Opportunities and Challenges of the department.
9. Research facilities and research output in the form of publications and patents.


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10. Computer, internet and library facilities available.
11. Mentoring system, introduction of Remedial classes, Bridge courses, guidance for NET/SET and competitive examinations.
12. Skill development and personality development programmes.
13. Generation of funds and optimum utilization.
14. Evaluation methods adopted for internal and external examinations.
15. Future plans of the department.

METHODOLOGY - NAAC has not given any specific methodology or guidelines for conducting Academic and Administrative audit but its general process is that IQAC committee of the institution prepares the plan to conduct academic and administrative audit. IQAC asks the departments to provide the data given or prepared by IQAC committee. The data should be related to all the aspects as per NAAC criteria. The data include the academic, curricular, co-curricular and extra-curricular activities conducted by the departments, achievements of the faculty and students in seminar, conferences and workshop at different level. The departments should provide and keep all the related information which are given in the format in NAAC under departmental evaluation section. After the preparation and collection of all the necessary documents IQAC should make the documentation properly and asks the academic and administrative committee for assessment. The periodicity is not fix some of the HIE's done it on regular basis and other are done it once in three years or five years. But the internal audit yearly and external audit once in three years or five years is good activity. The peer team or accessing agencies evaluate the institute as per its preparation and gives the outcome of the assessment to the institute.

CONCLUSION - Thus academic and administrative audit plays a vital role in the evaluation process of the institution. These aspects are very important as per NAAC has given the different gradation on which the institute's status are calculated and, on the gradation, further things are dependent. All the UGC and RUSA's funding system are depend on the gradation of institutions status. For that academic and administrative audit is very important to achieve certain desired aims.

In this way academic and administrative audit evaluate the institute and give their suggestions for further improvement. These are the scientific and systematic process by which the quality of academic process in the institution are reviewed. These parameters ensure



quality assurance and helps to enhance the the quality of academic and overall the procedures in HEI's.

Works Cited:

- Gupta B.L.(2011), Academic audit, Concept publishing, New Delhi.
- Rajendra, M(2007), Academic assessment ignoring administrative audit is farce accreditation in education in India, Atlantic publication, New Delhi.
- Manual for self-study affiliated college/constituent colleges.new-17dec19.pdf.
- NAAC_AAA_Note.pdf
- www.google.com.

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Institutions

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The present work was carried out during the period of two year from Apr 2009 to March 2011. During this investigation 17 physico-chemical parameters were studied such as Atmospheric temperature, Water Temperature, pH, Dissolved Oxygen, Total Solids (TS, TDS and TSS), Total Alkalinity, Phosphate, Nitrate, Calcium, Magnesium, Sulphate, Chloride, Sodium, Potassium, Electric Conductivity, Turbidity and Salinity. Similarly, during this study some Water Heavy Metals were studied such as Copper (Cu), Zinc (Zn), Iron (Fe), Lead (Pb) and Cadmium (Cd).



Sudarshan S Pedge
Sunil D Ahirrao



The author works as Assis. Prof. in Zoology at Kai. Rasika Mahavidyalaya Deoni. He did his M.Sc in Zoology from Department of Zoology, Yeshwant Mahavidyalaya Nanded. He received Ph. D degree in Zoology in 2013 from S.R.T.M.U.Nanded (MS, India). His area of Interest in Fishery science, Microbiology, Bioinformatics, Hydrochemistry, Ecological Studies.

Hydrochemistry of Lonar Meteoritic Crater (MS) India



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Dr. Sunil D. Ahirrao
IQAC-COORDINATOR
Kai. Rasika Mahavidyalaya, Deoni
Tq. Deoni Dist. Latur

Pedge, Ahirrao

Sudarshan S. Pedge
Principal
Kai. Rasika Mahavidyalaya, Deoni
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